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Preliminary communication

PREPARATIONS, STRUCTURES AND REACTIONS OF MONO-ALKENE ALKYNE COMPLEXES OF RHODIUM(1): THE CRYSTAL STRUCTURES OF ACETYL-ACETONATO (ETHYLENE) (HEXAFLUOROBUT-2-YNE) RHODIUM(1) AND ACETYL-ACETONATO (CYCLO-OCTENE) (HEXAFLUOROBUT-2-YNE) RHODIUM(1)

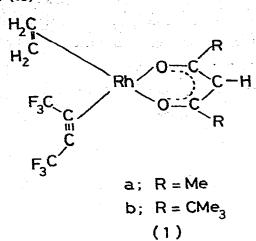
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## SUMMARY

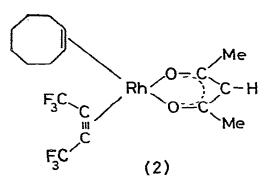
Mono-alkene alkyne complexes of rhodium(1) have been synthesised, and characterised by i.r., <sup>19</sup>F n.m.r., and crystallography. Their role as intermediates in the formation of cyclohexadienes is described.

The formation of cyclohexadienes by metal-catalysed cyclotrimerisations of two molecules of alkyne with one molecule of alkene proceed by both metallacyclopentene<sup>1</sup> and metallacyclopentadiene<sup>2,3</sup> intermediates. The metallacyclopentene intermediate complex is presumably derived from a mono-alkene alkyne precursor which then reacts with a further molecule of alkyne, but whereas diene alkyne complexes are known,<sup>4-6</sup> mono-alkene alkyne complexes have not hitherto been reported. We now describe the preparation, reactions, and structural characterisation of some mono-alkene alkyne complexes of rhodium(1). Treatment of the bis-ethylene complexes  $[Rh(acac)(C_2H_4)_2]$  and  $[Rh(dpm)(C_2H_4)_2]$ , (acac = MeCOCHCOMe; dpm = Me\_3CCOCHCOCMe\_3) in diethyl ether with hexafluorobut-2-yne at 195 K affords the novel alkene-alkyne complexes (1a) and (1b)



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The reaction of hexafluorobut-2-yne with [ $Rh(acac)(cis-cyclo-octene)_2$ ] at 195 K similarly yields the complex (2)

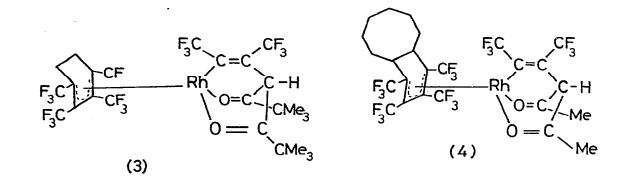


Each complex exhibits a strong band in its i.r. spectrum typical of an n-coordinated acetylene { (1a) 1976 cm<sup>-1</sup>; (1b) 1976 cm<sup>-1</sup>; (2) 1958 cm<sup>-1</sup>}, and a single resonance in the room temperature <sup>19</sup>F n.m.r. spectrum { (1a)  $\delta = -8.95$ ; (1b)  $\delta = -8.68$  p.p.m.in(CD<sub>3</sub>)<sub>2</sub>CO; (2)  $\delta = -9.91$  p.p.m. in CD<sub>2</sub>Cl<sub>2</sub>, all relative to  $\alpha, \alpha, \alpha$ -trifluorotoluene}.

The crystal structures of (1a) and (2) were determined using conventional methods with diffractometer data. (1a) Triclinic,  $\underline{PI}$ ,  $\underline{Z} = 2$ ,  $\underline{a} = 7.89(2)$ ,  $\underline{b} = 12.64(3)$ ,  $\underline{c} = 8.57(2)$ Å,  $\alpha = 95.9(2)^{\circ}$ ,  $\beta = 121.1(2)^{\circ}$ ,  $\check{\mathbf{b}} = 94.8(2)^{\circ}$ . Blocked full-matrix refinement gave R = 0.053 for 1228 observed reflections. (2) Monoclinic,  $\underline{B2}_1/\underline{c}$ ,  $\underline{Z} = 8$ ,  $\underline{a} = 14.16(3)$ ,  $\underline{b} = 17.89(3)$ ,  $\underline{c} = 15.71(3)$ Å,  $\beta = 97.5(2)^{\circ}$ . Block-diagonal least-squares refinement gave  $\underline{R} = 0.048$  for 1296 observed reflections.

The structural analyses verify that both complexes are monomeric, and have structures corresponding to those drawn for (1a) and (2). Important distances are: (1a), Rh-0 2.015(6), 2.033(7); Rh-C(alkyne) 2.042(11), 2.016(10); Rh-C(alkene) 2.171(17), 2.142(14); C=C 1.23(2); C=C 1.31(3)Å: (2), Rh-0 2.038(7), 2.017(8); Rh-C(alkyne) 2.059(10), 2.034(10); Rh-C(alkene) 2.196(11), 2.163(13); C=C 1.24(2); C=C 1.37(2)Å. Thus the geometries within the coordination spheres of the two complexes are very similar. We can now compare the structures of (1a) and (2) with those of acetylacetonatobis(ethylene)rhodium(1) and acetylacetonato(ethylene)(tetrafluoroethylene)rhodium(1), where replacement of one  $C_2H_4$  by  $C_2F_4$  causes a considerable weakening in the bonding between Rh and the remaining  $C_2H_4$ . The Rh-C( $C_2H_4$ ) distances in the bis(ethylene) complex are 2.117, 2.118(4)Å whereas those in the mixed complex are 2.190, 2.174(12)Å.<sup>7</sup> In (1a) and (2) these distances are again long, showing that the CF<sub>3</sub>CECCF<sub>3</sub> ligand has exerted a similar influence to CF<sub>2</sub>=CF<sub>2</sub>.

Treatment of the complexes (1b) and (2) with excess hexafluorobut-2yne at room temperature gives the cyclohexadiene complexes (3)<sup>8</sup> and (4) respectively. These reactions provide good evidence for the intermediacy



of mono-alkene alkyne complexes in the formation of cyclonexadiene com-8.9 plasses from bio-collegne shedium(i) complexes and dexariuorobut-2-yne.

The passage of carbon monoxide gas through an ether solution of (1b) results in the rapid displacement of both alkene and alkyne to give [ Rh(dpm)(CO) ]. However, the reactions of triphenyl-phosphine. -arsine, and -stibine with (ib) result in the displacement of ethylene only to give the complexes [ $9b(dpm)(C_{4}F_{5})L_{7}$ ],  $\{L = PPn_{2}, AsPn_{2}SoPn_{3}\}$ . This is consistent with the previously described weakening of the  $Rh = C_2 H_{\mu}$  bonds by hexafluorobut = 2-yne. These rhodium complexes are structurally related to the osmium complexes [ Os(CO)(NO)(RC=CR)(PPh<sub>2</sub>)<sub>2</sub>]  $[PF_{6}]$ , (R = H, Ph, CO<sub>2</sub>Me) for which there is evidence for rotation about the osmium- alkyne bond.<sup>10</sup> The room temperature <sup>19</sup>F n.m.r. spectrum of the complex [Rh(dpm)( $C_4F_6$ )(PPh<sub>3</sub>)<sub>2</sub>] exhibits one signal at -12.04 p.p.m.  $(J_{phr} = 2Hz$ , relative to internal  $\alpha, \alpha, \alpha$ -trifluorotoluene) which at 215 K splits into two signals at -12.08 p.p.m. (d, 3F, J<sub>pF</sub> = 16.8Hz) and -11.29 p.p.m. (s,3F). However, the absence of phosphorus-fluorine coupling in the room temperature <sup>19</sup>F n.m.r. spectrum of [ $Rh(dpm)(C_4F_6)(PPh_3)_2$ ], together with the observation that addition of excess triphenylphosphine raises the coalescence point in the low temperature spectrum from 254 K to 263 K, suggests that the temperature dependent  $^{19}$ F n.m.r. spectrum is a consequence of phosphine dissociation and not alkyne rotation in this complex.

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